

Mpemba effect

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The Mpemba effect has intrigued scientists for decades.



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About Mpemba effect:

- The Mpemba effect, named after Tanzanian student **Erasto Mpemba in 1969**, defies intuition as it reveals that **hot water can freeze faster than cold water under similar conditions**.
- Aristotle, Francis Bacon, and René Descartes had noticed similar effects centuries ago, but it gained scientific attention after Mpemba highlighted it in 1969.
- Despite early observations, researchers have recently conducted **numerous experiments** to delve into the causes of this intriguing phenomenon.

Proposed Explanations:

Microbubbles and Convection:

- Microbubbles left suspended in boiled water may promote convection, transferring heat faster as the water cools.
 - Convection is enhanced in warmer water due to its lower density.

Evaporation and Heat Transfer:

- Warmer water evaporates more, and evaporation is inherently endothermic (absorbing heat), contributing to faster heat loss.
 - It is similar to how sweat cools the skin.
- Accelerated heat transfer in warmer water aids the freezing process.

Frost as an Insulator:

• Presence of frost in cold water may act as an insulator, slowing the loss of heat and influencing freezing time.



Chemical Factors:

- The presence of compounds like calcium carbonate, precipitated by boiling and then dissolving.
 - o Dissolved compounds can act as impurities, affecting the freezing point

Complex Interplay of Mechanisms:

- The Mpemba effect continues to captivate scientists due to its intricate interplay of physical mechanisms.
- Researchers are yet to reach a consensus on a single explanation, indicating the complexity of factors involved.